You are planning a birthday party for your niece and need to make at least 4 gallons of Kool-Aid, which you would like to cool down to $32{ }^{\circ} \mathrm{F}\left(0^{\circ} \mathrm{C}\right)$ before the party begins. Unfortunately, your refrigerator is already so full of treats that you know there will be no room for the Kool-Aid. So, with a sudden flash of insight, you decide to start with 4 gallons of the coldest tap water you can get, which you determine is $50^{\circ} \mathrm{F}\left(10^{\circ} \mathrm{C}\right)$, and then cool it down with a 1 -quart chunk of ice you already have in your freezer. The owner's manual for your refrigerator states that when the freezer setting is on high, the temperature is $-20^{\circ} \mathrm{C}$. (To solve this you may need to consult your text or discussion group leader for some additional basic quantities once you identify them).
a) will your plan work?
b) if not, what is the final temperature achieved?

