

finish Concept Q 54.html

$$\#6 \quad \Delta T = 0 \Rightarrow \Delta U = 0 \Rightarrow \Delta U = Q - W \Rightarrow Q = W$$

$\#7$ again $\Delta T = 0$, and here $W < 0 \Rightarrow$ if $Q = W$ then $Q < 0$

$$\#8 \quad \Delta T = 0 \Rightarrow \Delta U = 0 \Rightarrow Q_{\text{tot}} = W_{\text{tot}} > 0$$

example

$P_a = 1 \text{ atm}$ piston (initial) $P_b = P_a = 1 \text{ atm}$ (final)

$\text{CO}_2 \quad n = 0.250 \text{ mol}$ $T_b = 127.0^\circ\text{C}$

$T_a = 27.0^\circ\text{C}$

a) PV diagram

b) How much work is done by the gas? $a \rightarrow b$

$$W = p \Delta V = p(V_2 - V_1) = nR \Delta T = (0.250 \text{ mol}) \left(8.3145 \frac{\text{J}}{\text{mol} \cdot \text{K}} \right) (100.0 \text{ K}) = 208 \text{ J}$$

c) On what is this work done? Piston

d) What is ΔU ? $\Delta U = n C_v \Delta T$ see Eqⁿ 3.9 and its discussion

$$\Delta U = (0.250 \text{ mol}) \left(28.93 \frac{\text{J}}{\text{mol} \cdot \text{K}} \right) (100.0 \text{ K}) = 723 \text{ J}$$

$\hookrightarrow C_v/R = 3.48$ for CO_2 Table 2.3

e) How much heat was supplied?

$$Q = \Delta U + W = 931 \text{ J}$$

f) What would W have been if $P_a = P_b = 0.50 \text{ atm}$?

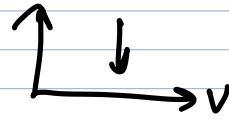
if ΔT unchanged, then in this case W unchanged

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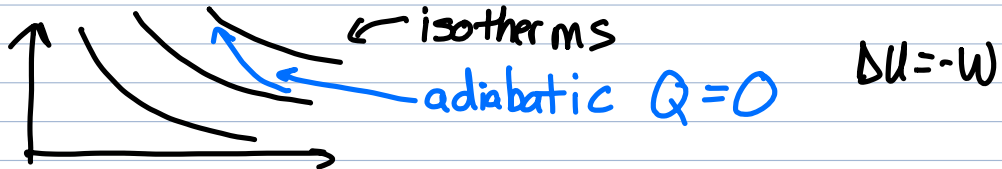
$$U \sim \frac{3}{2} nRT = \frac{3}{2} NkT \Rightarrow B \text{ smallest}$$

more PV fun!

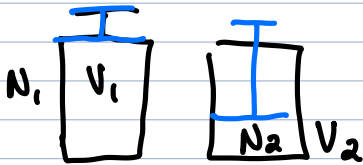
a) Your house gets cold at night.



b) Pump your bike tire



work is $0 \Rightarrow \Delta U > 0$



$$T_2 = T_1 \left(\frac{V_1}{V_2} \right)^{\gamma-1}$$

see p.130 ...

$$\text{where } \gamma = \frac{C_p}{C_v} \text{ and } C_p = C_v + R$$

$$\text{monatomic } \gamma = \frac{(C_v + R)/\frac{3}{2}R}{\frac{3}{2}R} = \left(\frac{\frac{3}{2}R + \frac{3}{2}R}{\frac{3}{2}R} \right) = \frac{5}{3} = 1.67$$

$$\text{diatomic } \gamma = \frac{(C_v + R)/\frac{5}{2}R}{\frac{5}{2}R} = \left(\frac{\frac{5}{2}R + \frac{5}{2}R}{\frac{5}{2}R} \right) = \frac{7}{5} = 1.4$$

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$$\textcircled{B} \Delta U = Q - W \rightarrow Q = \Delta U + W \text{ and } W > 0$$

moreover, total work done on piston is zero

$$W_{\text{by gas}} = -W_{\text{by gravity}} = - - mgh = mgh$$

Porsche Boxster, this time assuming adiabatic



P, V, T.

$$W_{0 \rightarrow 1} = \frac{1}{\gamma-1} (P_0 V_0 - P_1 V_1) \quad \text{p.131}$$

$$\text{where } T_0 V_0^{\gamma-1} = \text{constant} = T_1 V_1^{\gamma-1}$$

Eqn 3.14

$$\text{and thus } \frac{P_0 V_0}{Nk} V_0^{\gamma-1} = \frac{P_1 V_1}{Nk} V_1^{\gamma-1} \Rightarrow P_0 V_0^\gamma = P_1 V_1^\gamma \Rightarrow P_1 = P_0 \left(\frac{V_0}{V_1} \right)^\gamma$$

$$\rightarrow W_{0 \rightarrow 1} = \frac{1}{\gamma-1} (P_0 V_0 - P_0 \left(\frac{V_0}{V_1}\right)^\gamma V_1) = \frac{1}{1.4-1} (1.01 \cdot 10^5 \text{ Pa} \cdot 2687 \text{ cm}^3 - 1.01 \cdot 10^5 \text{ Pa} \left(\frac{11.3}{1}\right)^{1.4} \frac{2687 \text{ cm}^3}{11.3})$$

$$= -1111 \text{ J/stroke} \times \frac{7200 \text{ rpm}}{60 \text{ sec/min}} \cdot \frac{1 \text{ hp}}{746 \text{ W}} = -178 \text{ hp @ } 7200 \text{ rpm}$$